

A Triple Umpolung Sequence For the Preparation of Highly Substituted Indanes.

David C. Harrowven* and Rory Browne

Department of Chemistry, The University, Southampton, S017 1BJ.

Abstract: The paper describes a new method for the diastereoselective construction of highly substituted indanes. The key step involves a triple umpolung sequence in which a 2-lithio-2-vinyl-1,3-dithiane is first united with an aryl halide to form a ketenedithioacetal. Transmetalation of the aryl halide with an alkyl lithium reagent then generates an alkyl halide and a 2-indanyl-2-lithio-1,3-dithiane. Finally, these combine to complete the sequence. Copyright © 1996 Elsevier Science Ltd

We recently disclosed an approach to the *Podophyllum* lignans in which a Michael initiated ring closure (MIRC type II) featured as the key step (Figure 1).¹ This success lead us to consider how an analogous cyclopentannulation sequence might be accomplished. The problem did not appear to be a trivial one as it required the generation of an aryl cation equivalent in the presence of a lithiated dithiane. Reversing the polarisation of the system however, provided a potential solution (Figure 2).² For union of an *o*-bromobenzaldehyde and a 2-lithio-2-vinyl-1,3-dithiane would generate a ketenedithioacetal and, though unprecedented, we believed that an intramolecular cyclisation could then be induced by transmetalation of the halide.³

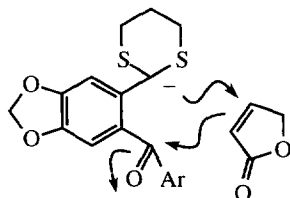


Figure 1

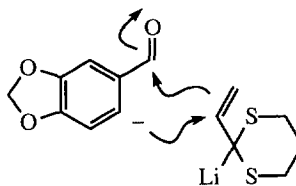
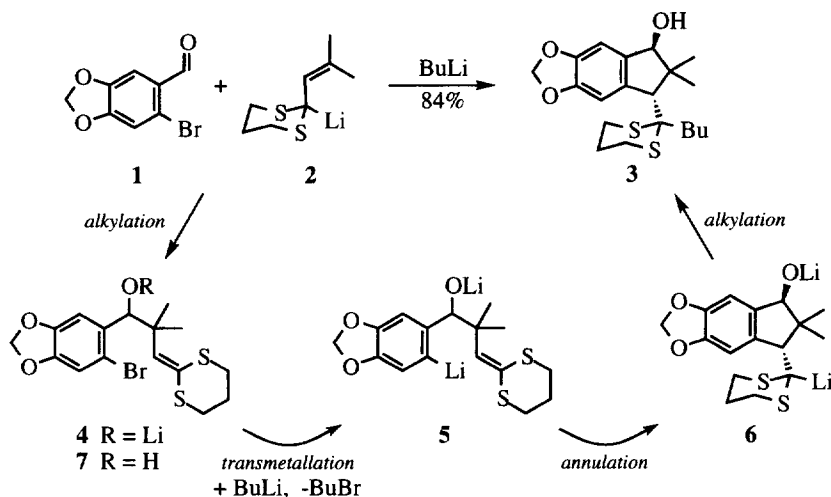
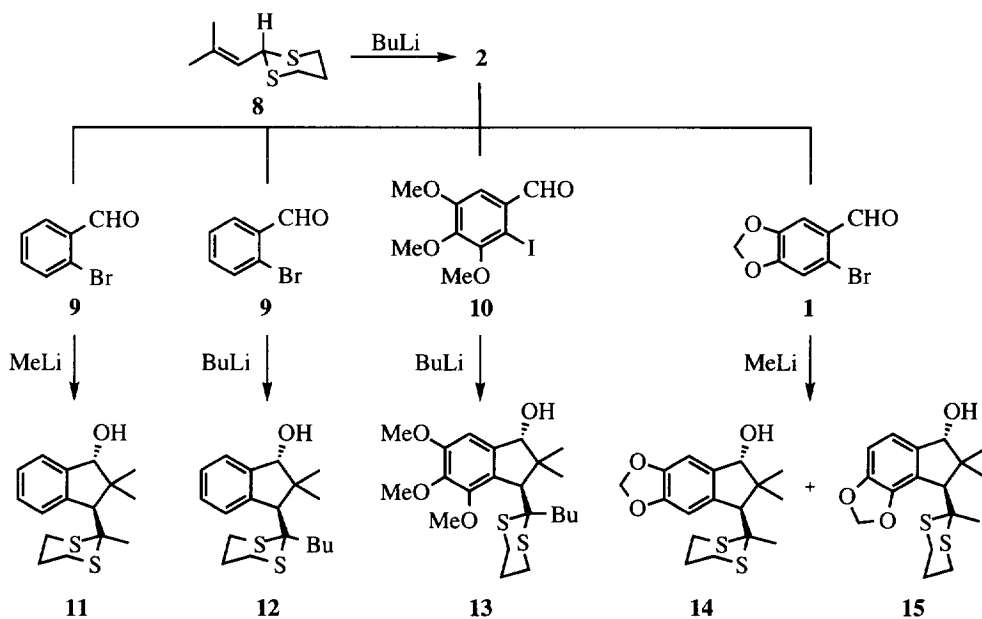


Figure 2

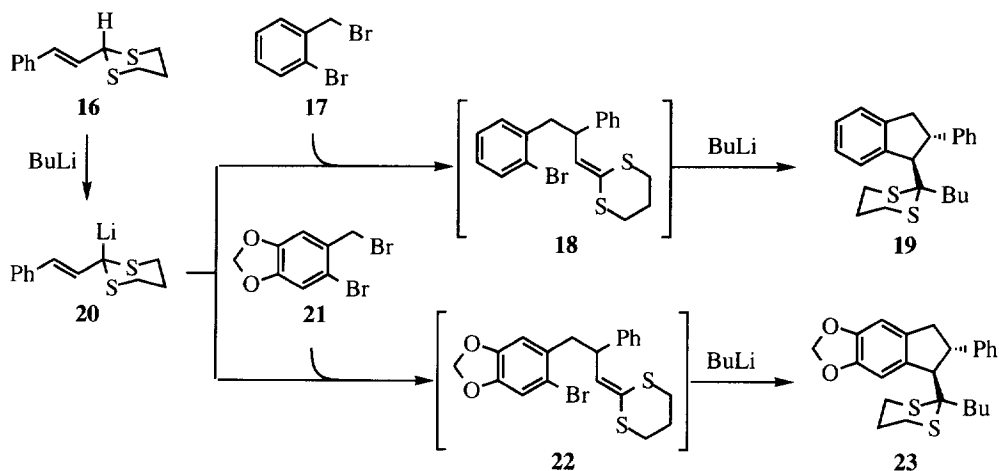
This general strategy was soon realised. Sequential addition of 6-bromopiperonal **1** and butyllithium to the lithiated 2-vinyl-1,3-dithiane **2**, for example, provided the indane **3** in 84% yield and as a single diastereoisomer (Scheme 1). Remarkably, in the one pot, three new carbon to carbon bonds had been established; each in an umpolung fashion.⁴



The indanes **11** to **15** have also been prepared from the lithiodithiane **2** using the above procedure (Scheme 2). These experiments demonstrated that the reaction was i) effective with aryl bromides and aryl iodides; ii) tolerant of ethereal functions on the aromatic ring; iii) *trans*-1,3-diastereoselective and iv) more efficient with butyllithium than with methyllithium. One unexpected result occurred when **2** was combined with 6-bromopiperonal **1** and then treated with methyllithium. This provided the anticipated indane **14** together with significant quantities of its regioisomer **15**; presumably reflecting the relative efficiency of transmetallation reactions mediated by MeLi and BuLi.⁵

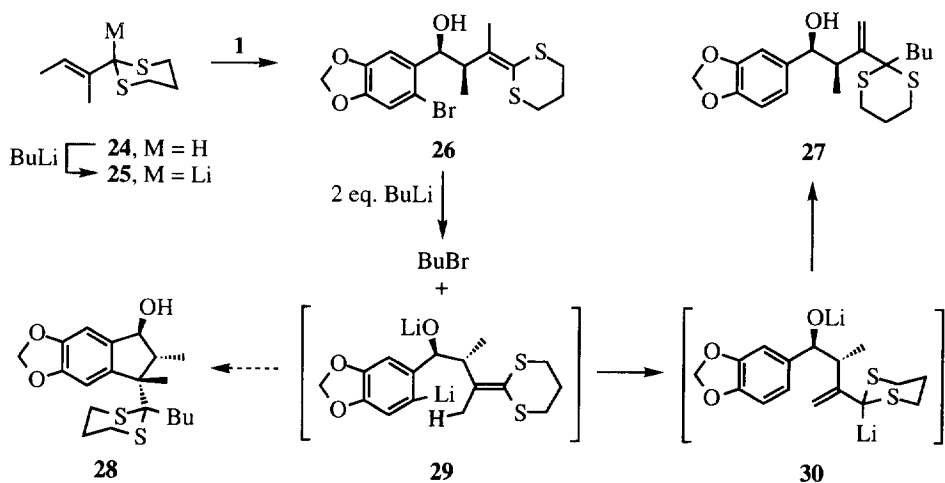


We have also shown the reaction to be *trans*-1,2-diastereoselective through use of the 2-lithio-1,3-dithiane **20** derived from *trans*-cinnamaldehyde. Thus, addition of 2-bromobenzyl bromide **17** to **20** first gave the ketenedithioacetal **18**, which was transformed into the indane **19** on exposure to BuLi. Similarly the benzodioxole **21** provided access to the indanodioxole **23** via ketenedithioacetal **22**.



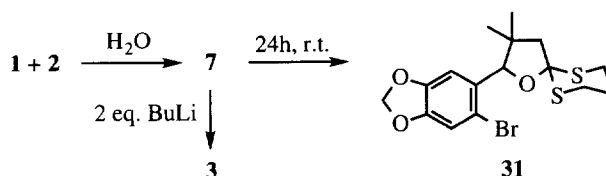
Scheme 3

Interestingly, the presence of a proton on the γ -carbon of the intermediate ketenedithioacetals had no discernible effect on the yield of the above reactions.⁶ By way of contrast, when the lithiated dithiane **25** was employed in this sequence the major product attained was the 2-vinyl-1,3-dithiane **27** rather than the indane **28**. Thus it would seem that intermediates akin to **29** will effect deprotonation when it is kinetically favourable to do so (Scheme 4).



Scheme 4

Finally, we should mention that the intermediate ketenedithioacetals can be intercepted and that these too are responsive to cyclisation upon treatment with two equivalents of an alkyllithium reagent. That the γ -hydroxy derivatives are prone to isomerisation on standing to the corresponding tetrahydrofuran detracts from the appeal of this course (*e.g.* Scheme 5).⁷



Scheme 5

In conclusion, we have developed a new cyclopentannulation reaction for the synthesis of highly substituted indanes. The procedure adopted involves a triple umpolung sequence and features the unprecedented addition of an aryllithium to a ketenedithioacetal. The high yields, diastereoselectivity, atom efficiency and easy of experimental are noteworthy.

EXPERIMENTAL

General Remarks.

Melting points were obtained using a Mel-Temp (II) apparatus and are uncorrected. UV spectra were recorded on a Pye Unicam SP800 spectrometer. Maxima and inflections (*inf*) are reported as λ_{max} followed in parentheses by the extinction coefficient ϵ ($\text{dm}^3\text{mol}^{-1}\text{cm}^{-1}$). IR spectra were recorded on a Perkin Elmer 1600 series Fourier transform infrared spectrometer using NaCl cells. Maxima are reported as ν_{max} followed by the signal intensity (described using the abbreviations *s*, strong; *m*, medium; *w*, weak; *v*, very; *br*, broad). ¹H n.m.r. spectra were recorded on a Bruker AC250 (250MHz) or a Jeol GX270 (270MHz) spectrometer. Chemical shifts (δ_{H}) are reported as values in parts per million relative to tetramethylsilane (δ 0.00) or residual CHCl_3 (δ 7.27). Multiplicities are described using the abbreviations *s*, singlet; *d*, doublet; *t*, triplet; *q*, quartet; *m*, multiplet; *br*, broad; *app.*, apparent. ¹³C spectra were recorded on a Bruker AC250 (63MHz) or a Jeol GX270 (68MHz) spectrometer. Chemical shifts (δ_{C}) are reported as values in parts per million relative to tetramethylsilane (δ 0.00) or residual CHCl_3 (δ 77.2). Multiplicities refer to the signals in the off-resonance spectra, as determined by DEPT 135° and DEPT 90° experiments, and are described using the abbreviations *s*, singlet; *d*, doublet; *t*, triplet; *q*, quartet. Two dimensional and n.O.e. experiments were recorded on a Jeol GX270 spectrometer. Mass spectra were recorded on a variety of instruments. Signals are reported as values in atomic mass units and are followed in parentheses by the peak intensity relative to the base peak (100%).

All reactions were magnetically stirred and conducted under a nitrogen atmosphere using flame dried glassware. Thin layer chromatography, using Macherey-Nagel Alugram Sil G/UV₂₅₄ precoated aluminium foil plates of layer thickness 0.25mm, was used to monitor reactions. Compounds were visualised firstly by UV irradiation then by heating plates exposed to solutions of potassium permanganate in aq. sodium carbonate. Column chromatography was performed on Sorbsil 60 silica (230-400 mesh), slurry packed and run under low pressure. THF was dried and degassed by refluxing over sodium wire using benzophenone ketyl as indicator. All reagents used were purchased from Lancaster Synthesis Ltd. or The Aldrich Chemical Company Ltd. and used as supplied.

trans-7-(2-butyl-1,3-dithian-2-yl)-6,7-dihydro-6,6-dimethyl-5*H*-indeno[5,6-*d*]-1,3-dioxol-5-ol 3

A hexane solution of *n*-butyllithium (1.6M, 9.7mL) was added over 10 min to a cooled (-20°C), THF solution (100mL) of 2-(2-methylprop-1-en-1-yl)-1,3-dithiane (2.45g, 14.1mmol). The resulting solution was warmed to 5°C over 2.5 h, then cooled to -50°C. A THF solution (40mL) of 6-bromopiperonal (3.23g, 14.1mmol) was then added over 15 min. After a further 30 min a hexane solution of *n*-butyllithium (1.6M, 9.7mL) added over 10 min. The resulting solution was allowed to warm to ambient temperature over 4 h then partitioned between ether (200mL) and water (200mL). The aqueous layer was extracted with further ether (2 x 100mL) and the combined organics washed with brine (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 10 to 50% ether in petrol) to yield the title compound (4.50g, 11.8mmol, 84%) as a white foam; **m.p.** 48-52°C; **FT-IR** (neat) ν_{\max} 3420brm, 2960s, 2770w, 1475s, 1275s, 1040s, 910s and 730s cm⁻¹; **UV** (CHCl₃) λ_{\max} (ϵ) 242 (6630), 246 (6230), 246 (6210) and 295 (5970) nm; **¹H NMR** (270MHz, CDCl₃) δ_{H} 7.14 (1H, s, ArH), 6.82 (1H, s, ArH), 5.98 (1H, d, *J* 1.4Hz, OCHHO), 5.96 (1H, d, *J* 1.4Hz, OCHHO), 5.09 (1H, d, *J* 7.8Hz, CHOH), 3.31 (1H, s, CHCS), 2.96-2.86 (2H, m, 2xSCH_{ax}H_{eq}), 2.83-2.68 (2H, m, 2xSCH_{ax}H_{eq}), 2.22 (1H, m, CHHCH₂CH₂CH₃), 2.14 (1H, d, *J* 7.8Hz, OH), 2.14-1.77 (3H, m, SCH₂CH₂CH₂SCCHH), 1.60 (2H, m, CH₂CH₂CH₃), 1.53 (3H, s, CCH₃), 1.36 (2H, sextet, *J* 7.6Hz, CH₂CH₃), 1.00 (3H, s, CCH₃) and 0.94 (3H, t, *J* 7.6Hz, CH₂CH₃) ppm (these assignments were confirmed by a ¹H-¹H COSY experiment); **n.O.e.** Irradiation of the signal at δ_{H} 1.53 (CCH₃) caused an n.O.e. enhancement at δ_{H} 5.09 (CHOH) of 14% while irradiation of the signal at δ_{H} 1.00 (CCH₃) caused an n.O.e. enhancement at δ_{H} 3.31 (CHCS) of 10%; **¹³C NMR** (68MHz, CDCl₃) δ_{C} 147.1 (s, C(Ar)), 146.4 (s, C(Ar)), 138.9 (s, C(Ar)), 134.3 (s, C(Ar)), 109.4 (d, CH(Ar)), 103.9 (d, CH(Ar)), 100.9 (t, OCH₂O), 81.9 (d, CHOH), 58.3 (d, CHCS), 58.2 (s, SCS), 50.5 (s, C(CH₃)₂), 36.7 (t, CH₂CS), 27.3 (t, CH₂CH₂CH₃), 26.4 (t, CH₂S), 25.9 (t, CH₂S), 25.2 (q, CCH₃), 24.3 (t, CH₂CH₂S), 23.8 (q, CCH₃), 22.8 (t, CH₂CH₃) and 14.0 (q, CH₂CH₃) ppm; **m/z** (EI) Found: 362.1423 (70%, [M-H₂O]⁺); C₂₀H₂₆S₂O₂ requires 362.1408, 288 (7), 254 (16), 241 (38), 227 (16), 213 (39), 199 (55), 188 (28) and 175 (100, [C₃H₆S₂CBu]⁺). (The intermediate ketenedithioacetal **7** could be intercepted and exhibited ¹H NMR (270MHz, CDCl₃) δ_{H} 7.02 (1H, s, ArH), 6.94 (1H, s, ArH), 6.02 (s, =CH), 5.96 (2H, s, OCH₂O), 5.20 (1H, d, *J* 3.2Hz, CHOH), 2.94 - 2.82 (4H, m, 2xSCH₂), 2.41 (1H, d, *J* 3.2Hz, OH), 2.14 - 2.04 (2H, m, SCH₂CH₂), 1.26 (3H, s, CH₃) and 1.19 (3H, s, CH₃) ppm. This sample rapidly isomerised on standing to the tetrahydrofuran **31** which exhibited ¹H NMR (270MHz, CDCl₃) δ_{H} 7.13 (1H, s, ArH), 7.00 (1H, s, ArH), 6.00 (2H, s, OCH₂O), 5.29 (1H, s, ArCH), 3.53 (1H, app. td, *J* 13.6, 2.9Hz, SCH_{ax}H_{eq}CH₂CH₂S), 3.42 (1H, app. td, *J* 13.6, 2.9Hz, SCH₂CH₂CH_{ax}H_{eq}S), 2.78 (1H, app. dt, *J* 13.6, 2.3Hz, SCH_{ax}H_{eq}CH₂CH₂S), 2.68 (1H, app. dt, *J* 13.6, 2.3Hz, SCH₂CH₂CH_{ax}H_{eq}S), 2.15 (2H, s, (CH₃)₂CCH₂C), 2.12-1.83 (2H, m, SCH₂CH₂CH₂S), 1.25 (3H, s, CH₃) and 0.88 (3H, s, CH₃) ppm.)

trans-3-(2-methyl-1,3-dithian-2-yl)-2,3-dihydro-2,2-dimethyl-1*H*-inden-1-ol 11

A hexane solution of *n*-butyllithium (1.6M, 4.8mL) was added over 5 min to a cooled (-20°C), THF solution (50mL) of 2-(2-methylprop-1-en-1-yl)-1,3-dithiane (1.22g, 7.05mmol). The resulting solution was warmed to ambient temperature over 3 h, then cooled to -78°C. A THF solution (10mL) of 2-bromobenzaldehyde (1.31g, 7.08mmol) was then added over 2 min. After a further 45 min a hexane solution of *n*-methylithium (1.4M, 5.5mL) was added. The resulting solution was allowed to warm to 0°C over 2 h then maintained at that temperature for 24h. The whole was partitioned between ether (200mL) and water (200mL), the aqueous layer was extracted with ether (100mL) and the combined organics were washed with brine (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 5 to 30% ether in petrol) to yield the title compound (1.49g, 5.07mmol, 72%) as colourless oil; **FT-IR** (neat) ν_{\max} 3400brm, 3070w, 3035w, 2970s, 2930s, 2900s, 1605w, 1475m, 1455m, 1420m, 1055s and 755s cm⁻¹; **UV** (CHCl₃) λ_{\max} (ϵ) 320inf

(210), 314inf (250), 309inf (260), 304inf (280), 292 (290), 274 (740), 264 (1130), 252 (1480) and 242 (1920) nm; **¹H NMR** (270MHz, CDCl₃) δ_H 7.80 (1H, d, *J* 8.2, ArH), 7.38-7.18 (3H, m, 3xArH), 4.91 (1H, brs, CHOH), 3.50 (1H, s, CHCS), 3.06-2.95 (2H, m, 2xSCH_{ax}H_{eq}), 2.87-2.76 (2H, m, 2xSCH_{ax}H_{eq}), 2.10-1.80 (3H, m, SCH₂CH₂ & OH), 1.80 (3H, s, SCCH₃), 1.43 (3H, s, CH₃) and 1.14 (3H, s, CH₃) ppm; **¹³C NMR** (68MHz, CDCl₃) δ_c 144.7 (s, C(Ar)), 141.1 (s, C(Ar)), 128.9 (d, CH(Ar)), 127.2 (d, CH(Ar)), 126.5 (d, CH(Ar)), 123.8 (d, CH(Ar)), 82.4 (d, CHOH), 60.4 (d, CHCS), 52.7 (s, SCS), 50.0 (s, C(CH₃)₂), 26.7 (t, CH₂S), 26.6 (t, CH₂S), 26.0 (q, SCCH₃), 25.0 (q, CCH₃), 24.9 (t, CH₂CH₂S) and 23.5 (q, CCH₃) ppm; **m/z** (EI) Found: 276.1044 (2%, [M-H₂O]⁺); C₁₆H₂₀S₂ requires 276.1007, 187 (9) and 133 (100, [C₃H₆S₂CMe]⁺).

***trans*-3-(2-butyl-1,3-dithian-2-yl)-2,3-dihydro-2,2-dimethyl-1*H*-inden-1-ol 12**

A hexane solution of *n*-butyllithium (1.6M, 4.6mL) was added over 10 min to a cooled (-20°C), THF solution (50mL) of 2-(2-methylprop-1-en-1-yl)-1,3-dithiane (1.23g, 7.07mmol). The resulting solution was warmed to ambient temperature over 3 h, then cooled to -78°C. A THF solution (10mL) of 2-bromobenzaldehyde (1.30g, 7.03mmol) was then added over 15 min. After a further 45 min a hexane solution of *n*-butyllithium (1.6M, 4.6mL) was added, dropwise over 10 min. The resulting solution was allowed to warm to ambient temperature over 6h then partitioned between ether (200mL) and water (200mL). The aqueous layer was extracted with further ether (2 x 100mL) and the combined organics were dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 5 to 30% ether in petrol) to yield the title compound (2.11g, 6.28mmol, 89%) as a pale yellow oil: **FT-IR** (neat) ν_{max} 3415brm, 3070w, 2955s, 2920s, 2870, 1920w, 1605w, 1460m, 1215m, 1050m and 755s cm⁻¹; **¹H NMR** (250MHz, CDCl₃) δ_H 7.61 (1H, d, *J* 7.4, ArH), 7.35-7.18 (3H, m, 3xArH), 5.23 (1H, brs, CHOH), 3.42 (1H, s, CHCS), 2.95-2.80 (2H, m, 2xSCH_{ax}H_{eq}), 2.73-2.58 (2H, m, 2xSCH_{ax}H_{eq}), 2.25 (1H, m, CHHCH₂CH₂CH₃), 2.08-1.73 (3H, m, SCH₂CH₂CH₂S-CCHH), 1.69-1.15 (3H, m, CH₂CH₂CH₃ & OH), 1.57 (3H, s, CCH₃), 1.36 (2H, sextet, *J* 7.2Hz, CH₂CH₃), 0.98 (3H, s, CCH₃) and 0.97 (3H, t, *J* 7.2Hz, CH₂CH₃) ppm; **¹³C NMR** (63MHz, CDCl₃) δ_c 145.5 (s, C(Ar)), 141.4 (s, C(Ar)), 129.0 (d, CH(Ar)), 127.7 (d, CH(Ar)), 126.7 (d, CH(Ar)), 123.4 (d, CH(Ar)), 82.4 (d, CHOH), 58.6 (d, CHCS), 58.2 (s, SCS), 50.2 (s, C(CH₃)₂), 36.9 (t, CH₂CS), 27.5 (t, CH₂CH₂CH₃), 26.5 (t, CH₂S), 26.0 (t, CH₂S), 25.1 (q, CCH₃), 24.5 (t, CH₂CH₂S), 23.8 (q, CCH₃), 22.9 (t, CH₂CH₃) and 14.1 (q, CH₂CH₃) ppm; **m/z** (CI, NH₃) Found: 337.1660 (7%, [MH]⁺); C₁₉H₂₉S₂O requires 337.1660, 319 (12, [MH-H₂O]⁺), 229 (6) and 175 (100, [C₃H₆S₂CBu]⁺).

***trans*-3-(2-methyl-1,3-dithian-2-yl)-2,3-dihydro-2,2-dimethyl-4,5,6-trimethoxy-1*H*-inden-1-ol 13**

A hexane solution of *n*-butyllithium (1.6M, 4.8mL) was added over 5 min to a cooled (-20°C), THF solution (50mL) of 2-(2-methylprop-1-en-1-yl)-1,3-dithiane (1.22g, 7.05mmol). The resulting solution was warmed to ambient temperature over 3 h, then cooled to -60°C. A THF solution (10mL) of 2-iodo-3,4,5-trimethoxybenzaldehyde (2.27g, 7.05mmol) was then added over 5 min. After a further 45 min a hexane solution of methylolithium (1.4M, 5.5mL) added. The resulting solution was allowed to warm to ambient temperature over 3 h then maintained at that temperature for 15h. The whole was partitioned between ether (200mL) and water (200mL), the aqueous layer was extracted with ether (2x100mL) and the combined organics were washed with ammonium chloride (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 5 to 50% ether in petrol) to yield the title compound (1.84g, 4.79mmol, 68%) as a colourless oil; **FT-IR** (CHCl₃) ν_{max} 3420brm, 2935s, 1602m, 1465s, 1410s, 1335s, 1235m, 1195m, 1115s, 1050m and 755s cm⁻¹; **UV** (CHCl₃) λ_{max} (ε) 280 (4600) and 242 (12000) nm; **¹H NMR** (250MHz, CDCl₃) δ_H 6.71 (1H, s, ArH), 5.46 (1H, brs, CHOH), 3.87 (3H, s, OCH₃), 3.84 (6H, s, 2xOCH₃), 3.20 (1H, s, CHCS), 3.11 (1H, ddd, *J* 14.4, 11.8, 2.5Hz, SCH_{ax}H_{eq}), 2.96 (1H, ddd, *J* 14.4, 11.8, 2.5Hz, SCH_{ax}H_{eq}), 2.75 (1H, app.dt, *J* 14.4,

4.0Hz, SCH_{ax}H_{eq}), 2.61 (1H, app.dt, *J* 14.4, 4.0Hz, SCH_{ax}H_{eq}), 2.20 (1H, brs, OH), 2.09 (1H, m, SCH₂CH_{ax}H_{eq}), 1.86 (3H, s, SCCH₃), 1.80 (1H, m, SCH₂CH_{ax}H_{eq}), 1.74 (3H, s, CH₃) and 0.86 (3H, s, CH₃) ppm; ¹³C NMR (68MHz, CDCl₃) δ_c 154.2 (s, C(Ar)), 151.6 (s, C(Ar)), 141.7 (s, C(Ar)), 140.9 (s, C(Ar)), 125.9 (s, C(Ar)), 102.4 (d, CH(Ar)), 81.3 (d, CHOH), 61.2 (q, OCH₃), 60.8 (q, OCH₃), 60.5 (q, OCH₃), 55.9 (d, CHCS), 55.5 (s, SCS), 51.5 (s, C(CH₃)₂), 28.5 (q, SCCH₃), 27.5 (t, CH₂S), 26.4 (t, CH₂S), 25.5 (t, CH₂CH₂S), 25.5 (q, CCH₃), and 23.4 (q, CCH₃) ppm; ^{m/z} (CI, NH₃) Found: 385.1507 (13%, [MH]⁺); C₁₉H₂₉O₄S₂ requires 385.1507, 367 (3, [MH-H₂O]⁺), 251 (18, [M-C₃H₆S₂CMe]⁺) and 133 (100, [C₃H₆S₂CMe]⁺).

trans-7-(2-methyl-1,3-dithian-2-yl)-6,7-dihydro-6,6-dimethyl-5*H*-indeno[5,6-*d*]-1,3-dioxol-5-ol 14 and

trans-8-(2-methyl-1,3-dithian-2-yl)-7,8-dihydro-7,7-dimethyl-6*H*-indeno[4,5-*d*]-1,3-dioxol-6-ol 15

A hexane solution of *n*-butyllithium (1.6M, 9.4mL) was added over 5 min to a cooled (-20°C), THF solution (50mL) of 2-(2-methylprop-1-en-1-yl)-1,3-dithiane (2.45g, 14.1mmol). The resulting solution was warmed to ambient temperature over 4 h, then cooled to -78°C. A THF solution (20mL) of 6-bromopiperonal (3.23g, 14.1mmol) was then added over 2 min. After a further 15 min a hexane solution of *n*-methylithium (1.4M, 13mL) added over 2 min. The resulting solution was warm to ambient temperature, stirred for 8 h then partitioned between ether (200mL) and water (200mL). The aqueous layer was extracted with a aliquot of ether (100mL) and the combined organics were washed with brine (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 10 to 50% ether in petrol) to yield firstly *trans*-8-(2-methyl-1,3-dithian-2-yl)-7,8-dihydro-7,7-dimethyl-6*H*-indeno[4,5-*d*]-1,3-dioxol-6-ol (1.52g, 4.51mmol, 32%) as a white solid; **m.p.** 152-155°C; **CHN** Found: C 59.9%, H 6.5%; C₁₇H₂₂S₂O₃ requires C 60.3%, H 6.6%; **FT-IR** (CHCl₃) ν_{max} 3420brm, 2915s, 2865s, 2770m, 1870w, 1465s, 1365s, 1300s, 1280m, 990m and 945s cm⁻¹; **UV** (CHCl₃) λ_{max} (ε) 295 (4810), 251 (4340) and 247 (4510) nm; **¹H NMR** (250MHz, CDCl₃) δ_H 7.36 (1H, s, Ar*H*), 6.93 (1H, s, Ar*H*), 5.97 (2H, s, OCH₂O), 4.78 (1H, brd, *J* 6.9Hz, CHOH), 3.39 (1H, s, CHCS), 3.03 (2H, ddd, *J* 14.5, 9.7, 3.4Hz, 2xSCH_{ax}H_{eq}), 2.83 (2H, ddd, *J* 14.5, 6.4, 3.6Hz, 2xSCH_{ax}H_{eq}), 2.03 (1H, m, SCH₂CH_{ax}H_{eq}), 1.94 (1H, m, SCH₂CH_{ax}H_{eq}), 1.82 (3H, s, CH₃), 1.70 (1H, brd, *J* 6.9Hz, OH), 1.43 (3H, s, CH₃) and 1.18 (3H, s, CH₃) ppm; ¹³C NMR (63MHz, CDCl₃) δ_c 147.1 (s, C(Ar)), 146.8 (s, C(Ar)), 138.4 (s, C(Ar)), 134.6 (s, C(Ar)), 109.9 (d, CH(Ar)), 104.4 (d, CH(Ar)), 101.1 (t, OCH₂O), 82.8 (d, CHOH), 60.6 (d, CHCS), 53.2 (s, SCS), 50.7 (s, C(CH₃)₂), 27.0 (t, CH₂S), 26.8 (t, CH₂S), 26.3 (q, SCCH₃), 25.2 (q, CCH₃), 24.9 (t, CH₂CH₂S) and 23.9 (q, CCH₃) ppm; ^{m/z} (EI) Found: 320.0869 (8%, [M-H₂O]⁺); C₁₇H₂₀S₂O₂ requires 320.0904, 214 (10), 199 (10), 187 (6, [M-H₂O-C₃H₆S₂CMe]⁺) and 133 (100, [C₃H₆S₂CMe]⁺) then *trans*-7-(2-methyl-1,3-dithian-2-yl)-6,7-dihydro-6,6-dimethyl-5*H*-indeno[5,6-*d*]-1,3-dioxol-5-ol (2.47g, 7.31mmol, 52%) as a colourless oil; **FT-IR** (CHCl₃) ν_{max} 3425brm, 2965s, 2895s, 2830w, 2780w, 1840w, 1605w, 1500m, 1460s, 1245s, 1045s, 910s and 730s cm⁻¹; **UV** (CHCl₃) λ_{max} (ε) 291 (4900), 288 (4910), 251inf (5220) and 249 (5240) nm; **¹H NMR** (250MHz, CDCl₃) δ_H 6.82 (1H, dd, *J* 7.8, 1.1Hz, Ar*H*), 6.76 (1H, d, *J* 7.8Hz, Ar*H*), 5.93 (1H, d, *J* 1.4Hz, OCHHO), 5.88 (1H, d, *J* 1.4Hz, OCHHO), 5.25 (1H, brs, CHOH), 3.30 (1H, s, CHCS), 3.04 (1H, ddd, *J* 14.4, 10.4, 3.1Hz, SCH_{ax}H_{eq}), 2.93 (1H, ddd, *J* 14.4, 10.5, 3.1Hz, SCH_{ax}H_{eq}), 2.84 (1H, ddd, *J* 14.4, 5.9, 3.4Hz, SCH_{ax}H_{eq}), 2.71 (1H, ddd, *J* 14.4, 5.9, 3.3Hz, SCH_{ax}H_{eq}), 2.10-1.80 (3H, m, SCH₂CH₂ & OH), 1.80 (3H, s, SCCH₃), 1.64 (3H, s, CH₃) and 0.85 (3H, s, CH₃) ppm; ¹³C NMR (68MHz, CDCl₃) δ_c 146.7 (s, C(Ar)), 144.5 (s, C(Ar)), 140.6 (s, C(Ar)), 122.2 (s, C(Ar)), 116.1 (d, CH(Ar)), 107.8 (d, CH(Ar)), 100.1 (t, OCH₂O), 80.5 (d, CHOH), 59.8 (d, CHCS), 54.4 (s, SCS), 50.7 (s, C(CH₃)₂), 27.6 (q, SCCH₃), 27.1 (t, CH₂S), 26.2 (t, CH₂S), 25.2 (q, CCH₃), 24.9 (t, CH₂CH₂S) and 23.2 (q, CCH₃) ppm (the structural assignment was confirmed by ¹³C-¹H COSY and ¹³C-¹H COLOC experiments); ^{m/z} (EI) Found: 231.1006 (12%, [M-C₃H₇S₂]⁺); C₁₄H₁₅O₃ requires 231.1021, 189 (5) and 133 (100, [C₃H₆S₂CMe]⁺).

trans-3-(2-butyl-1,3-dithian-2-yl)-2,3-dihydro-2-phenyl-1*H*-indene 18

A hexane solution of *n*-butyllithium (1.6M, 3.1mL) was added over 3 min to a cooled (-78°C), THF solution (50mL) of 2-(2-phenyleth-1-en-1-yl)-1,3-dithiane (1.00g, 4.5mmol) resulting in an orange coloured solution. After 30 min 2-bromobenzyl bromide (1.13g, 4.5mmol), as a solution in THF (10mL), was added over 5 min and the solution stirred for 30 min. A hexane solution of *n*-butyllithium (1.6M, 3.1mL) was then added over 5 min. The resulting solution was stirred for 30 min at -78°C then warmed to ambient temperature over 1 h. Water (6mL) was added and the whole partitioned between ether (200mL) and water (200mL). The organic layer was dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 2 to 10% ether in petrol) to yield the title compound (1.31g, 3.56mmol, 79%) as a white solid; **m.p.** 131-133°C (ether / petrol); **FT-IR** (CHCl₃) ν_{\max} 3025w, 2955m, 2940m, 2870m, 2845m, 1960w, 1750w, 1610w, 1490m, 1460m and 910m cm⁻¹; **UV** (CHCl₃) λ_{\max} (ϵ) 239 (2620), 256 (1420), 261 (1430), 269 (1440) and 276 (1190) nm; **¹H NMR** (270MHz, CDCl₃) δ_{H} 7.66 (1H, d, *J* 7.6Hz, Ar*H*), 7.41-7.21 (8H, m, 8xAr*H*), 4.15 (1H, brd, *J* 8.9Hz, CHPh), 3.97 (1H, brs, CHCHPh), 3.73 (1H, dd, *J* 16.8, 8.9Hz, CHHCHPh), 3.05 (1H, brd, *J* 16.8Hz, CHHCHPh), 2.94 (2H, m, 2xSCH_{ax}H_{eq}), 2.78 (2H, m, 2xSCH_{ax}H_{eq}), 2.00 (3H, m, SCH₂CH₂ & SCCHH), 1.80-1.55 (3H, m, SCCHHCH₂), 1.36 (2H, sextet, *J* 7.3Hz, CH₂CH₃) and 0.98 (3H, t, *J* 7.3Hz, CH₃) ppm (these assignments were confirmed by ¹H-¹H COSY and ¹³C-¹H COSY experiments); **¹³C NMR** (68MHz, CDCl₃) δ_{C} 149.1 (s, C(Ph)), 145.3 (s, C(Ar)), 140.7 (s, C(Ar)), 128.4 (d, 2xCH(Ph)), 127.9 (d, CH(Ar)), 127.6 (d, CH(Ar)), 126.2 (d, 2xCH(Ph)), 125.9 (d, CH(Ph)), 125.8 (d, CH(Ar)), 124.1 (d, CH(Ar)), 60.0 (d, CHCHPh), 59.6 (s, SCS), 46.4 (d, CHPh), 41.0 (t, CH₂CHPh), 34.9 (t, CH₂CS), 26.8 (t, CH₂CH₂CH₃), 25.9 (t, CH₂S), 25.8 (t, CH₂S), 24.6 (t, CH₂CH₂S), 23.0 (t, CH₂CH₃) and 14.0 (q, CH₂CH₃) ppm (these assignments were confirmed by a ¹³C-¹H COSY experiment); **m/z** (EI) Found: 368.1677 (4%, [M]⁺); C₂₃H₂₈S₂ requires 368.1666. 293 (3, [M-C₃H₆SH]⁺), 261 (4, [M-SC₃H₆SH]⁺), 193 (9, [M-C₃H₆S₂CBu]⁺), 175 (100, [C₃H₆S₂CBu]⁺), 115 (14) and 91 (20).

The intermediate ketenedithioacetal **22**, a pale yellow oil, could be intercepted and exhibited **UV** (CHCl₃) λ_{\max} (ϵ) 295 (4600), 261 inf (6180), 256 inf (6730), 252 (7620), 248 (8340) and 243 (9060) nm; **¹H NMR** (250MHz, CDCl₃) δ_{H} 7.43 - 7.18 (5H, m, C₆H₅), 6.98 (1H, s, Ar*H*), 6.58 (1H, s, Ar*H*), 6.19 (1H, d, *J* 9.9Hz, =CH), 5.92 (2H, s, OCHHO), 4.26 (1H, ddd, *J* 9.9, 9.2, 6.1Hz, PhCH), 3.07 (1H, dd, *J* 13.6, 6.1Hz, ArCHH), 2.94 (1H, dd, *J* 13.6, 9.2Hz, ArCHH), 2.85 - 2.55 (4H, m, 2xSCH₂) and 2.07 (2H, m, SCH₂CH₂CH₂) ppm; **¹³C NMR** (63MHz, CDCl₃) δ_{C} 146.9 (s, C(Ph)), 146.7 (s, C(Ar)), 143.1 (s, C(Ar)), 135.6 (d, =CH), 131.9 (s, C(Ar)), 128.6 (d, 2xCH(Ph)), 127.8 (s, C(Ar)), 127.5 (d, 2xCH(Ph)), 126.5 (d, CH(Ph)), 115.2 (s, =CS₂), 112.5 (d, CH(Ar)), 111.1 (d, CH(Ar)), 101.5 (t, OCH₂O), 45.6 (d, CHPh), 42.7 (t, CH₂CHPh), 30.3 (t, CH₂S), 29.6 (t, CH₂S) and 25.1 (t, CH₂CH₂S) ppm; **m/z** sample suffered hydrolysis prior to analysis.

trans-5-(2-butyl-1,3-dithian-2-yl)-6,7-dihydro-6-phenyl-5*H*-indenol[5,6-*d*]-1,3-dioxole 23

A hexane solution of *n*-butyllithium (1.6M, 3.1mL) was added over 3 min to a cooled (-78°C), THF solution (50mL) of 2-(2-phenyleth-1-en-1-yl)-1,3-dithiane (1.00g, 4.5mmol) resulting in deep red solution. After 10 min a THF solution (10mL) of 6-bromopiperonyl bromide (1.10g, 3.74mmol) was added and the solution stirred for 30 min. A hexane solution of *n*-butyllithium (1.6M, 3.1mL) was then added over 5 min, the resulting solution was warmed to 0°C over 30 min, then partitioned between ether (200mL) and water (200mL). The aqueous layer was extracted with further aliquots of ether (2 x 100mL) then the combined organics were washed with brine (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 5 to 15% ether in petrol) to yield the title compound (1.31g, 3.18mmol, 85%) as a white solid; **m.p.** 128-129°C (ether / petrol); **FT-IR** (CHCl₃) ν_{\max} 2955m, 2925m, 2870m, 1605w, 1500m, 1475s, 1295m, 1245m, 1040m, 940m, 755m and 700m cm⁻¹; **UV** (CHCl₃) λ_{\max} (ϵ) 240 (7820), 243 inf (7030) and 299 (11000) nm; **¹H NMR**

(250MHz, CDCl₃) δ_H 7.25-7.11 (5H, m, C₆H₅), 7.16 (1H, s, ArH), 7.04 (1H, s, ArH), 5.99 (1H, d, *J* 1.3Hz, OCHHO), 5.96 (1H, d, *J* 1.3Hz, OCHHO), 4.01 (1H, d, *J* 8.9Hz, CHHCHPh), 3.73 (1H, s, CHCHPh), 3.51 (1H, dd, *J* 16.6, 8.9Hz, CHPh), 2.98-2.63 (4H, m, SCH₂CH₂CH₂S), 2.83 (1H, d, *J* 16.6Hz, CHHCHPh), 1.97-1.80 (3H, m, SCH₂CH₂CH₂SCCHH), 1.72-1.40 (3H, m, CHHCH₂CH₂CH₃), 1.28 (3H, sextet, *J* 7.2Hz, CH₂CH₃) and 0.89 (3H, t, *J* 7.2Hz, CH₃) ppm (these assignments were confirmed by a ¹H-¹H COSY experiment); ¹³C NMR (63MHz, CDCl₃) δ_c 149.2 (s, C(Ph)), 147.6 (s, C(Ar)), 146.3 (s, C(Ar)), 138.3 (s, C(Ar)), 133.3 (s, C(Ar)), 128.5 (d, 2xCH(Ph)), 126.3 (d, 2xCH(Ph)), 126.0 (d, CH(Ph)), 108.1 (d, CH(Ar)), 104.2 (d, CH(Ar)), 101.0 (t, OCH₂O), 59.7 (d, CHCHPh), 59.6 (s, SCS), 47.2 (d, CHPh), 40.8 (t, CH₂CHPh), 34.9 (t, CH₂CS), 26.8 (t, CH₂CH₂CH₃), 25.9 (t, CH₂S), 25.8 (t, CH₂S), 24.6 (t, CH₂CH₂S), 23.1 (t, CH₂CH₃) and 14.0 (q, CH₂CH₃) ppm; ^{m/z} (EI) 237 (13%, [M-C₃H₆S₂CBu]⁺), 207 (7), 175 (100, [C₃H₆S₂CBu]⁺) and 136 (38).

6-bromo-5-(3-(1,3-dithian-2-ylidene)-1S^{*}-hydroxy-2S^{*}-methylbut-1-yl)benzo[1,3]dioxole 26

A hexane solution of *n*-butyllithium (1.6M, 3.8mL) was added over 5 min to a cooled (-20°C), THF solution (50mL) of 2-(but-2-en-2-yl)-1,3-dithiane (1.00g, 5.75mmol). The resulting solution was warmed to ambient temperature over 3 h, then cooled to -60°C. A THF solution (10mL) of 6-bromopiperonal (1.30g, 5.68mmol) was then added over 5 min. The resulting solution was allowed to warm to ambient temperature over 3 h then partitioned between ether (200mL) and water (200mL). The aqueous layer was extracted with ether (2x100mL) and the combined organics were dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 5 to 50% ether in petrol) to yield the title compound (2.11g, 5.24mmol, 92%) as a white solid; **m.p.** 156-157°C; **CHN Found:** C 47.5%, H 4.7%, Br 19.7%, S 15.9; C₁₆H₁₉S₂O₃Br requires C 47.7%, H 4.8%, Br 19.8%, S 15.9; **FT-IR** (CHCl₃) ν_{max} 3415brm, 2990w, 2910m, 2850m, 2770m, 2070w, 1860w, 1625m, 1500m, 1485m, 1465m, 1375m, 1110s, 995s and 965s cm⁻¹; **UV** (CHCl₃) λ_{max} (ε) 319 (420), 290inf (4500), 257inf (10400), 254 (10500) and 245 (9970) nm; **¹H NMR** (250MHz, CDCl₃) δ_H 7.09 (1H, s, ArH), 6.95 (1H, s, ArH), 5.97 (1H, d, *J* 1.4Hz, OCHHO), 5.95 (1H, d, *J* 1.4Hz, OCHHO), 4.95 (1H, d, *J* 8.7Hz, CHOH), 3.71 (1H, dq, *J* 8.7, 6.7Hz, CHCH₃), 2.90-2.63 (4H, m, 2xSCH₂), 2.12-1.85 (2H, m, SCH₂CH₂), 1.76 (3H, s, =CCH₂) and 1.20 (3H, d, *J* 6.7Hz, CHCH₃) ppm; **¹³C NMR** (68MHz, CDCl₃) δ_c 147.6 (s, C(Ar)), 147.3 (s, C(Ar)), 140.5 (s, C(Ar)), 135.1 (SCS), 121.5 (s, C(Ar)), 113.8 (s, C(Ar)), 112.2 (d, CH(Ar)), 108.6 (d, CH(Ar)), 101.7 (t, OCH₂O), 75.5 (d, CHOH), 43.2 (d, CHCH₃), 30.3 (t, CH₂S), 29.9 (t, CH₂S), 24.9 (t, CH₂CH₂S), 16.2 (q, CHCH₃) and 15.1 (q, CCH₃) ppm; **^{m/z} (EI) Found:** 404.0004 (9%, [M(⁸¹Br)]⁺) and 401.9921 (8%, [M(⁷⁹Br)]⁺); C₁₆H₁₉BrO₃S₂ requires 403.9939 (⁸¹Br) and 401.9959 (⁷⁹Br); 386 (25, [M(⁸¹Br)-H₂O]⁺), 384 (23, [M(⁷⁹Br)-H₂O]⁺), 339 (18), 337 (17), 305 (17, [M-Br-H₂O]⁺), 242 (30), 240 (28), 230 (29), 228 (29), 173 (68), 125 (58), 106 (51) and 40 (100).

5-(2-(2-butyl-1,3-dithian-2-yl)-4S^{*}-hydroxy-3S^{*}-methylbut-1-en-4-yl)benzo[1,3]dioxole 27

A hexane solution of *n*-butyllithium (1.6M, 3.4mL) was added over 2 min to a cooled (-78°C), THF solution (50mL) of **26** (1.00g, 2.48mmol). The resulting solution was warmed to 0°C over 30 min, stirred for a further 4h then partitioned between ether (200mL) and water (200mL). The organics were washed with brine (200mL), dried (MgSO₄), filtered, concentrated *in vacuo*, and purified by column chromatography (gradient elution, 10 to 25% ether in petrol) to yield the title compound (395mg, 1.04mmol, 42%) as a colourless oil; **FT-IR** (neat) ν_{max} 3445brm, 2965m, 2930m, 2900m, 2865m, 1500m, 1485m, 1440m, 1235m, 1040s and 930m cm⁻¹; **¹H NMR** (250MHz, CDCl₃) δ_H 6.98 (1H, d, *J* 1.5Hz, ArH), 6.90 (1H, dd, *J* 8.0, 1.5Hz, ArH), 6.81 (1H, d, *J* 8.0, ArH), 5.98 (2H, s, OCH₂O), 5.85 (1H, brs, =CHH), 5.65 (1H, d, *J* 0.7Hz, =CHH), 4.84 (1H, brd, *J* 2.4Hz, CHOH), 2.97-2.65 (5H, m, 2xSCH₂ & CHCH₃), 2.25 (1H, s, OH), 2.13-1.82 (4H, m, SCH₂CH₂ & CH₂CH₂CH₂CH₃), 1.40-1.22 (4H,

m, $\text{CH}_2\text{CH}_2\text{CH}_3$), 0.98 (3H, d, J 6.7Hz, CHCH_3) and 0.89 (3H, t, J 7.1Hz, CH_2CH_3) ppm (these assignments were confirmed by a ^1H - ^1H COSY experiment); ^{13}C NMR (68MHz, CDCl_3) δ_{c} 149.8 (s, =CH), 147.4 (s, C(Ar)), 146.3 (s, C(Ar)), 137.6 (s, C(Ar)), 119.0 (d, CH(Ar)), 117.8 (t, = CH_2), 107.8 (d, CH(Ar)), 106.7 (d, CH(Ar)), 100.8 (t, OCH_2O), 75.1 (d, CHOH), 60.0 (s, SCS), 41.8 (d, CHCH_3), 39.5 ($\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 27.9 (t, CH_2S), 27.6 (t, CH_2S), 26.0 (t, $\text{CH}_2\text{CH}_2\text{CH}_3$), 25.4 (t, $\text{CH}_2\text{CH}_2\text{S}$), 22.8 (t, CH_2CH_3), 15.1 (q, CHCH_3) and 13.9 (q, CCH_3) ppm (these assignments were confirmed by a ^{13}C - ^1H COSY experiment); m/z (ES+) 363 (100%, $[\text{MH}-\text{H}_2\text{O}]^+$)

ACKNOWLEDGEMENTS

The majority of this work was conducted at The University of Nottingham. The authors wish to thank that institution for a teaching fellowship (to DCH), Prifysgol Cymru, Bangor for an M.Sc. studentship (to RB) and Professor G. Pattenden for his interest in this work.

REFERENCES AND NOTES

1. a. Harrowven, D.C. *Tetrahedron Lett.*, **1991**, 32, 3735; b. Harrowven, D.C.; Dennison, S.T. *Tetrahedron Lett.*, **1993**, 34, 3323; c. Harrowven, D.C. *Tetrahedron*, **1993**, 49, 9039. See also d. McCombie, S.W.; Togat, J.R.; Metz, W.A.; Nazareno, D.; Puar, M.S. *Tetrahedron*, **1993**, 49, 8073; e. Kamal, A.; Daveshtalab, M.; Atchison, K.; Micetich, R.G. *Bioorg. Med. Chem. Lett.*, **1994**, 1513; f. Kamal, A.; Daveshtalab, M.; Micetich, R.G. *Tetrahedron Lett.*, **1994**, 35, 3879.
2. For overviews of dithiane chemistry and umpolung reactivity see a. Page, P.C.B.; van Niel, M.B.; Prodger, J.C. *Tetrahedron*, **1989**, 45, 7643; b. Cook, M.J. Six-membered Rings with More than One Oxygen or Sulfur Atom. In *Comprehensive Heterocyclic Chemistry*, Katritzky, A.R.; Rees, C.W. Eds., Pergamon: Oxford, Vol. 3, 1984, pp 943; c. Seebach, D. *Angew. Chem., Int. Ed. Engl.*, **1984**, 23, 639; d. Seebach, D. *Angew. Chem., Int. Ed. Engl.*, **1979**, 18, 239; e. Gröbel, B.T.; Seebach, D. *Synthesis*, **1977**, 357.
3. For reviews pertaining to the chemistry of ketenedithioacetals see a. Junjappa, H.; Ila, H.; Asokan; C.V. *Tetrahedron*, **1990**, 46, 5423; b. Kolb, M. *Synthesis*, **1990**, 171; c. Kolb, M. Ketene Thioacetals. In *The Chemistry of Ketenes, Allenes and Related Compounds*, Patai, S. Ed.; Wiley: Chichester, 1980, pp 669.
4. For related work from our group see a. Harrowven, D.C.; Browne, R. *Tetrahedron Lett.*, **1995**, 36, 2861; b. Harrowven, D.C.; Browne, R. *Tetrahedron Lett.*, **1994**, 35, 5301; c. Harrowven, D.C. *Tetrahedron Lett.*, **1993**, 34, 5653; d. Harrowven, D.C. *Tetrahedron Lett.*, **1992**, 33, 2879.
5. For a study on the relative efficiency of transmetallation reactions with organolithium reagents see a. Swenton, J.S.; Jackson, D.K.; Manning, M.J.; Raynolds, P.W. *J. Am. Chem. Soc.*, **1978**, 100, 6182; b. Ho, T.-L. *Tetrahedron*, **1985**, 41, 1.
6. a. Ziegler, F.E.; Fang, J.-M.; Tam, C.C. *J. Am. Chem. Soc.*, **1982**, 104, 7174; b. Lindgren, B.; Stjernlöf, P.; Trogen, L. *Acta Chem. Scand. Ser. B*, **1987**, 41, 180; c. Seebach, D.; Bürstinghaus, R.; Gröbel, B.-T.; Kolb, M. *Liebigs Ann. Chem.*, **1977**, 830; d. Murphy, W.S.; Wattanasin, S. *J. Chem. Soc., Perkin Trans. 1*, **1980**, 2678; e. Andersen, N.H.; Duffy, P.F.; Denniston, A.D.; Grotjahn, D.B. *Tetrahedron Lett.*, **1978**, 4315.
7. a. Corey, E.J.; Beames, D.J. *J. Am. Chem. Soc.*, **1973**, 95, 5829; b. Koch, S.S.; Chamberlin, A.R. *J. Org. Chem.*, **1993**, 58, 2725; c. Lee, T.-J.; Holtz, W.J.; Smith, R.L. *J. Org. Chem.*, **1982**, 47, 4750; d. Kozikowski, A.P.; Chen, Y.-Y. *J. Org. Chem.*, **1980**, 45, 2236; e. Suzuki, K.; Tomooka, K.; Katayama, E.; Matsumoto, T.; Tsuchihashi, G. *J. Am. Chem. Soc.*, **1986**, 108, 5221; f. Bellingham, R.; Jarowicki, K.; Kocienski, P.; Martin, V. *Synthesis*, **1996**, 285; g. Hatanaka, M. *Tetrahedron Lett.*, **1987**, 28, 83.